

Name: _____ Period _____

For each of the following elements, look up the following information from the periodic table.

Argon

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Chlorine

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Sodium

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Calcium

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Helium

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Bromine

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Potassium

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Magnesium

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Neon

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Fluorine

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Lithium

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Beryllium

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Iron

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Gold

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Oxygen

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Group Name: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____

Silicon

Symbol: _____
Metal Classification: _____
Atomic Number: _____
Average Atomic Mass: _____
Period Number: _____
Group Number: _____
Number Protons: _____
Number Electrons: _____
Number Neutrons: _____